

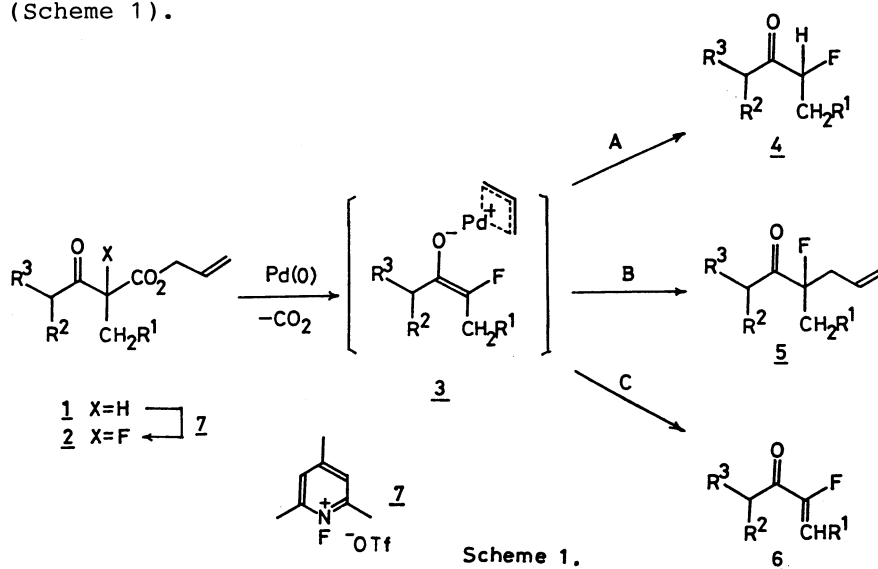
Short Effective Synthesis of  $\alpha$ -Fluoroketones by Palladium-Catalyzed  
Decarboxylation Reactions of Allyl  $\alpha$ -Fluoro- $\beta$ -keto Carboxylates

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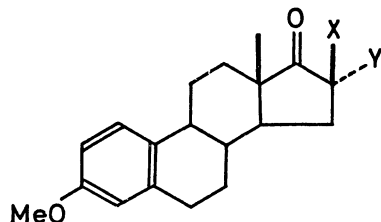
$\alpha$ -Fluoroketones,  $\alpha$ -fluoro- $\alpha$ -allylketones, and  $\alpha$ -fluoro- $\alpha,\beta$ -unsaturated ketones are synthesized by palladium-catalyzed decarboxylation reactions of allyl  $\alpha$ -fluoro- $\beta$ -keto carboxylates.

Although  $\alpha$ -fluoroketones occupy an important position for synthesis of various fluorinated compounds, only a few practical synthetic methods for  $\alpha$ -fluoroketones have been reported.<sup>1)</sup> Recently, Umemoto has developed N-fluoropyridinium triflates as mild fluorinating agents of active hydrogen compounds, such as  $\beta$ -keto esters.<sup>2)</sup> However, decarboxylation of  $\alpha$ -fluoro- $\beta$ -keto esters to the corresponding ketones are difficult owing to acyl fission reaction (retro-Claisen condensation) during hydrolysis. We have reported that allylic  $\beta$ -keto esters are decarboxylated easily using palladium catalysts to give ketones by decarboxylation-hydrogenolysis,<sup>3)</sup>  $\alpha$ -allylketones by decarboxylation-allylation,<sup>4)</sup> and  $\alpha,\beta$ -unsaturated ketones by decarboxylation-dehydrogenation.<sup>5)</sup> We wish to report here facile synthetic methods for the preparation of  $\alpha$ -fluoroketones,<sup>1b)</sup>  $\alpha$ -fluoro- $\alpha$ -allylketones,<sup>1c)</sup> and  $\alpha$ -fluoro- $\alpha,\beta$ -unsaturated ketones<sup>1d)</sup> from allyl  $\alpha$ -fluoro- $\beta$ -keto carboxylates based on the palladium-catalyzed decarboxylation reactions (Scheme 1).



Decarboxylation of allyl 2-fluorocyclododecanone carboxylate (**2a**)<sup>6)</sup> to 2-fluorocyclododecanone (**4a**) was carried out using formic acid in the presence of palladium catalyst (Method A). A mixture of HCO<sub>2</sub>H (88%, 0.14 ml) and Et<sub>3</sub>N (0.21 ml) in dioxane (5 ml) was added to a mixture of Pd<sub>2</sub>(dba)<sub>3</sub>CHCl<sub>3</sub> (0.019 mmol) and PPh<sub>3</sub> (0.019 mmol) in dioxane (10 ml) at room temperature under argon. To the mixture was added a solution of **2a** (0.75 mmol) in dioxane (5 ml). The reaction mixture was stirred for 24 hr at room temperature, poured into saturated NaHCO<sub>3</sub>, and extracted with ether. The usual workup and purification by column chromatography on SiO<sub>2</sub> gave **4a** in 80% yield: **4a** <sup>1</sup>H NMR (CDCl<sub>3</sub>, 90.0 MHz) δ 1.61-2.16 (m, 3H), 2.43-2.79 (m, 1H), 4.86 (ddd, J=49.0, 7.0, and 4.3 Hz, 1H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 22.5 MHz) δ 34.3, 95.5 (d, J<sub>CF</sub>=184.5 Hz), 208.5 (d, J<sub>CCF</sub>=21.4 Hz); <sup>19</sup>F NMR (CDCl<sub>3</sub>, 84.7 MHz, internal CFCl<sub>3</sub>) δ -188.67--189.74 (m); IR (neat) 1721 cm<sup>-1</sup>; HRMS Found 200.1558, Calcd for C<sub>12</sub>H<sub>21</sub>OF 200.1577.

This decarboxylation method was successfully applied to the synthesis of 16β-fluoroestrone methyl ether (**12**).<sup>8)</sup> The allyl α-fluoro-β-keto ester **11** was prepared from estrone methyl ether (**8**) in three steps and decarboxylated using HCO<sub>2</sub>H with Pd catalyst under reflux to give **12** in 93% yield. The stereochemistry at C-16 of **12** was determined by <sup>1</sup>H and <sup>19</sup>F NMR spectra.<sup>8b,10)</sup> Formation of the α-fluoro epimer at C-16 was not detected in this reaction. The selectivity is easily explained by the preferential α-side protonation of the enolate of the 17-keto steroid.



- |   |   |         |
|---|---|---------|
| <b>8.</b> X, Y=H  | ← | a (78%) |
| <b>9.</b> X, Y=H, CO <sub>2</sub> Me                                  | ← | b (94%) |
| <b>10.</b> X, Y=H, CO <sub>2</sub> CH <sub>2</sub> CH=CH <sub>2</sub> | ← | c (78%) |
| <b>11.</b> X, Y=F, CO <sub>2</sub> CH <sub>2</sub> CH=CH <sub>2</sub> | ← | d (93%) |
| <b>12.</b> X=F, Y=H   | ← |         |

a) (MeO)<sub>2</sub>CO, KH in THF reflux. b) Cl(n-Bu)<sub>2</sub>Sn-O-Sn(n-Bu)<sub>2</sub>OH (1mol%), HO-CH<sub>2</sub>-CH=CH<sub>2</sub> in toluene reflux.<sup>9)</sup> c) NaH, **7** in THF at 0 °C. d) HCO<sub>2</sub>H, Et<sub>3</sub>N,  $\frac{1}{2}$ Pd<sub>2</sub>(dba)<sub>3</sub>CHCl<sub>3</sub> (5 mol%), PPh<sub>3</sub> (2.5 mol%), in dioxane reflux.

When the reaction of **2a** was carried out without using formic acid, decarboxylation-allylation took place to give the α-allyl-α-fluoroketone **5a** (Method B). Thus, a solution of **2a** (0.56 mmol) in dry THF (10 ml) was added to a refluxing mixture of Pd<sub>2</sub>(dba)<sub>3</sub>CHCl<sub>3</sub> (0.014 mmol) and 1,2-bis(diphenylphosphino)ethane (dppe) (0.056 mmol) in dry THF (10 ml) under argon. The mixture was refluxed for 1 h. The usual workup and purification by chromatography on SiO<sub>2</sub> to give **5a** in 73% yield: **5a** <sup>1</sup>H NMR (CCl<sub>4</sub>, 60.0 MHz) δ 4.83-5.05 (m, 1H), 5.13 (s, 1H), 5.37-6.37 (m, 1H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 22.5 MHz) δ 34.5, 39.8 (d, J<sub>CCF</sub>=22.1 Hz), 102.4 (d, J<sub>CF</sub>=185.2 Hz), 118.8, 130.8 (d, J<sub>CCCF</sub>=3.6 Hz), 210.0 (d, J<sub>CCF</sub>=26.3 Hz); <sup>19</sup>F NMR (CDCl<sub>3</sub>, 84.7 MHz, internal CFCl<sub>3</sub>) δ -160.72--161.68 (m); IR (neat) 1719, 1472, 992, and 917 cm<sup>-1</sup>; HRMS Found 240.1906, Calcd for C<sub>15</sub>H<sub>25</sub>OF 240.1890.

Decarboxylation-dehydrogenation of **2a** to the α,β-unsaturated ketone **6a** was

also carried out in  $\text{CH}_3\text{CN}$  (Method C). A solution of **2a** (0.75 mmol) in dry  $\text{CH}_3\text{CN}$  (4 ml) was added to a refluxing mixture of  $\text{Pd}_2(\text{dba})_3\text{CHCl}_3$  (0.038 mmol) and  $\text{PPh}_3$  (0.038 mmol) in dry  $\text{CH}_3\text{CN}$  (3.5 ml) under argon, and the mixture was refluxed for 2 hr to give **6a** in 74% yield: **6a**  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 90.0 MHz)  $\delta$  2.16-2.72 (m, 4H), 6.02 (dt,  $J=35.5$  and 8.4 Hz, 1H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 22.5 MHz)  $\delta$  37.2 (d,  $J_{\text{CCCF}}=1.4$  Hz), 119.6 (d,  $J_{\text{CCF}}=12.4$  Hz), 156.3 (d,  $J_{\text{CF}}=262.5$  Hz), 196.6 (d,  $J_{\text{CCF}}=31.1$  Hz);  $^{19}\text{F}$  NMR ( $\text{CDCl}_3$ , 84.7 MHz, internal  $\text{CFCl}_3$ )  $\delta$  -127.20 (d,  $J=35.5$  Hz); IR (neat) 1703, 1650, 1467, and 1445  $\text{cm}^{-1}$ ; HRMS Found 198.1408, Calcd for  $\text{C}_{12}\text{H}_{19}\text{OF}$  198.1420.

The conversion of **2a** to **4a**, **5a**, and **6a** are considered to occur via the  $\pi$ -allylpalladium enolates of the  $\alpha$ -fluoroketones **3** which generate after oxidative addition of the ester **2** and subsequent decarboxylation. Similarly as shown in Table 1,  $\alpha$ -fluorocyclohexanones **5b** and **5c**, and  $\alpha$ -fluorocyclohexenones **6b** and **6c** were obtained from the corresponding esters **2b** and **2c** with high regioselectivity.<sup>10)</sup>

Table 1. Palladium-Catalyzed Decarboxylation Reactions of Allyl  $\alpha$ -Fluoro- $\beta$ -Keto Carboxylates to  $\alpha$ -Fluoroketones

Run	Ester	$\text{R}^1, \text{R}^2$	$\text{R}^3$	Method	Product	Yield/% <sup>a)</sup>
1	<b>2a</b>	$-(\text{CH}_2)_8-$	H	A	2-Fluorocyclododecanone ( <b>4a</b> )	80
2		$-(\text{CH}_2)_8-$	H	B	2-Allyl-2-fluorocyclododecanone ( <b>5a</b> )	73
3		$-(\text{CH}_2)_8-$	H	C	(E)-2-Fluoro-2-cyclododecenone ( <b>6a</b> )	74
4	<b>2b</b>	$-(\text{CH}_2)_2-$	H	B	2-Allyl-2-fluorocyclohexanone ( <b>5b</b> )	45
5		$-(\text{CH}_2)_2-$	H	C	2-Fluoro-2-cyclohexenone ( <b>6b</b> )	56
6	<b>2c</b>	$-(\text{CH}_2)_2-$	Me	B <sup>b)</sup>	2-Allyl-2-fluoro-6-methylcyclohexanone ( <b>5c</b> )	66
7		$-(\text{CH}_2)_2-$	Me	C	1-Fluoro-6-methyl-2-cyclohexenone ( <b>6c</b> )	61

a) Isolated yield. b)  $\text{PPh}_3$  (20 mol%) was used instead of dppe.

Since the reaction proceeds under mild conditions, the present methods would provide a useful synthetic method for the synthesis of fluorinated compounds. Further application of these methods to synthesis of biologically active compounds is in progress.

We thank Dr. Umemoto for helpful discussion on the fluorination method. This work was financially supported from the Ministry of Education, Science and Culture of Japan (No. 63750841), and the foundation, Hattori Hokokai.

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- 6) The  $\alpha$ -fluoro ester 2a was prepared in 84% yield from allyl cyclododecanone carboxylate (1a) by fluorination using 7:<sup>2,7</sup>) 2a <sup>1</sup>H NMR (CCl<sub>4</sub>, 60.0 MHz)  $\delta$  1.10-2.87 (m, 20H), 4.58 (d, J=5.5 Hz, 2H), 5.04-5.45 (m, 2H), 5.54-6.23 (m, 1H); <sup>19</sup>F NMR (CDCl<sub>3</sub>, 84.7 MHz, internal CFCl<sub>3</sub>) -166.29 (d, J=26.2 Hz); IR (neat) 1758 and 1731 cm<sup>-1</sup>; HRMS Found 284.1801, Calcd for C<sub>16</sub>H<sub>25</sub>O<sub>3</sub>F 284.1788.
- 7) The reagent 7 is commercially available from Wako Pure Chemical Industries, Ltd.
- 8) a) T. B. Patric and R. Mortezenia, *J. Org. Chem.*, 53, 5153 (1988); b) D. O. Kiesewetter, J. A. Katzenellenbogen, M. R. Kilbourn, and M. J. Welch, *ibid.*, 49, 4900 (1984).
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- 10) Spectral data. 5b <sup>1</sup>H NMR (CCl<sub>4</sub>, 60.0 MHz)  $\delta$  4.83-5.04 (m, 1H), 5.15 (s, 1H), 5.43-6.11 (m, 1H); <sup>19</sup>F NMR (CDCl<sub>3</sub>, 84.7 MHz, internal CFCl<sub>3</sub>)  $\delta$  -157.61--158.51 (m); IR (neat) 1728, 1432, 1127, and 923 cm<sup>-1</sup>; HRMS Found 156.0958, Calcd for C<sub>9</sub>H<sub>13</sub>OF 156.0951. 6b <sup>1</sup>H NMR (CCl<sub>4</sub>, 60.0 MHz)  $\delta$  6.31 (dt, J=14.8 and 4.0 Hz, 1H); <sup>19</sup>F NMR (CDCl<sub>3</sub>, 84.7 MHz, internal CFCl<sub>3</sub>)  $\delta$  -130.40 (d, J=14.8 Hz); IR (neat) 1693, 1180, 1146, 1112, and 893 cm<sup>-1</sup>. 5c <sup>1</sup>H NMR (CCl<sub>4</sub>, 60.0 MHz)  $\delta$  1.01 (d, J=6.5 Hz, 3H), 4.85-5.06 (m, 1H), 5.16 (s, 1H), 5.40-6.05 (m, 1H); <sup>19</sup>F NMR (CDCl<sub>3</sub>, 84.7 MHz, internal CFCl<sub>3</sub>)  $\delta$  -153.96--154.69 (m); IR (neat) 1726, 1453, 1128, and 981 cm<sup>-1</sup>; HRMS Found 170.1135, Calcd for C<sub>10</sub>H<sub>15</sub>OF 170.1107. 6c <sup>1</sup>H NMR (CCl<sub>4</sub>, 60.0 MHz)  $\delta$  1.14 (d, J=6.5 Hz, 3H), 6.21 (dt, J=15.0 and 4.0 Hz, 1H); <sup>19</sup>F NMR (CDCl<sub>3</sub>, 84.7 MHz, internal CFCl<sub>3</sub>)  $\delta$  -130.66 (d, J=15.0 Hz); IR (neat) 1691, 1453, 1346, 1195, 1180, 1005, 926, and 910 cm<sup>-1</sup>; HRMS Found 128.0604, Calcd for C<sub>7</sub>H<sub>9</sub>OF 128.0637. 11 <sup>1</sup>H NMR (CCl<sub>4</sub>, 60.0 MHz)  $\delta$  1.09 (s, 3H), 1.22-3.04 (m, 13H), 3.69 (s, 3H), 4.66 (d, J=5.5 Hz, 2H), 5.06-5.49 (m, 2H), 5.56-6.25 (m, 1H), 6.64 (d, J=7.0 Hz, 1H), 6.66 (d, J=9.5 Hz, 1H), 7.14 (d, J=9.5 Hz, 1H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 22.5 MHz)  $\delta$  55.11, 66.60, 95.28 (d, J<sub>CF</sub>=201.1 Hz), 111.58, 113.80, 119.20, 126.09, 130.75, 131.24, 137.28, 157.58, 167.37 (d, J<sub>CCF</sub>=26.3 Hz), 207.88 (d, J<sub>CCF</sub>=15.9 Hz); <sup>19</sup>F NMR (CDCl<sub>3</sub>, 84.7 MHz, internal CFCl<sub>3</sub>)  $\delta$  -161.23 (t, J=16.2 Hz); HRMS Found 386.1896, Calcd for C<sub>23</sub>H<sub>27</sub>O<sub>4</sub>F 386.1894. 12 <sup>1</sup>H NMR (CCl<sub>4</sub>, 60.0 MHz)  $\delta$  1.02 (s, 3H), 1.17-3.12 (m, 13H), 3.75 (s, 3H), 4.73 (dt, J=49.9 and 7.5 Hz, 1H), 6.66 (d, J=7.0 Hz, 1H), 6.68 (d, J=9.5 Hz, 1H), 7.16 (d, J=9.5 Hz, 1H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 22.5 MHz)  $\delta$  54.94, 91.40 (d, J<sub>CF</sub>=194.8 Hz), 111.43, 113.66, 125.92, 131.30, 137.22, 157.46, 212.63 (d, J<sub>CCF</sub>=11.6 Hz); <sup>19</sup>F NMR (CDCl<sub>3</sub>, 84.7 MHz, internal CFCl<sub>3</sub>)  $\delta$  -184.86 (dd, J=49.9 and 21.4 Hz).

(Received December 27, 1988)